ZINC CONCENTRATE
SAFETY DATA SHEET

SECTION 1. IDENTIFICATION

Product Identity: Red Dog Zinc Concentrate.

Trade Names and Synonyms: None.

Manufacturer:
Teck Alaska Incorporated
Red Dog Mine
P.O. Box 1230
Kotzebue, Alaska
99752
Emergency Telephone: (250) 364-4214

Supplier:
Teck Alaska Incorporated
Red Dog Mine
P.O. Box 1230
Kotzebue, Alaska
99752

Preparer:
Teck Metals Ltd.
Suite 3300 – 550 Burrard Street
Vancouver, British Columbia
V6C 0B3

Date of Last Review: June 12, 2015.

Date of Last Edit: June 12, 2015.

Product Use: Zinc concentrate is used in the production of zinc metal and zinc alloys.

SECTION 2. HAZARDS IDENTIFICATION

CLASSIFICATION:

Health		Physical	Environmental
Acute Toxicity (Oral, Inhalation)	– Does not meet criteria	Does not meet criteria for any Physical Hazard	**Aquatic Toxicity – Long Term – Category 3**
Skin Corrosion/Irritation	– Does not meet criteria		
Eye Damage/Eye Irritation	– Does not meet criteria		
Respiratory or Skin Sensitization	– Does not meet criteria		
Mutagenicity	– Does not meet criteria		
Carcinogenicity	**– Category 1A**		
Reproductive Toxicity	**– Category 1A**		
Specific Target Organ Toxicity:			
Acute Exposure	– Does not meet criteria		
Chronic Exposure	**– Category 1**		

LABEL:

Symbols: 

Signal Word: **DANGER**

Hazard Statements	Precautionary Statements:
DANGER! May cause cancer through inhalation of dust. May damage fertility or the unborn child. Causes damage to the respiratory system through prolonged or repeated exposure. Harmful to aquatic life with long lasting effects.	Obtain special instructions before use. Do not handle until all safety precautions have been read and understood. Do not breathe dust. Wear protective gloves, protective clothing, and eye protection. Wash hands thoroughly after handling. Do not eat, drink or smoke when using this product. Avoid release to the environment. If exposed or concerned or you feel unwell: Get medical advice/attention. Collect all spillage.

Emergency Overview: A dark brown, heavy, soil-like material that is not flammable or combustible under normal conditions of transport and storage. However, when heated strongly in air it will burn, releasing toxic and irritating sulphur dioxide gas as well as possible lead and zinc oxide fumes. Contact with strong acids will generate flammable and highly toxic hydrogen sulphide gas (H_2S). Inhalation or ingestion of concentrate dust may produce chronic health effects. Possible cancer hazard due to lead, cadmium and silica content. Possible reproductive hazard due to the lead content. SCBA and full protective clothing required for fire emergency response personnel.

Potential Health Effects: *Caution: The toxicological properties of this material have not been fully investigated. The information contained in this SDS is therefore based on information in the technical and scientific literature about the material's constituent compounds*.

Concentrate dust is irritating to the nose, throat and respiratory tract. Inhalation or ingestion of very high concentrations of concentrate dust may result in lead and cadmium absorption and possible lead intoxication. Symptoms include headache, nausea, vomiting, abdominal spasms, fatigue, sleep disturbances, weight loss, anemia and leg, arm, and joint pain. Prolonged exposure may also cause central nervous system damage (e.g., fatigue, headaches, tremors, hypertension), gastrointestinal disturbances, anemia, kidney dysfunction and possible reproductive effects. Pregnant women should be protected from excessive exposure to prevent lead crossing the placental barrier and causing infant neurological disorders. Lead and lead compounds are listed as an A3 Carcinogen (Confirmed Animal Carcinogen with Unknown Relevance to Humans) by the ACGIH. IARC has listed lead compounds as Group 2A Carcinogens (Probably Carcinogenic to Humans). Cadmium is classified as an A2 Carcinogen by the ACGIH and as a Group 1 Carcinogen by IARC. Silica is classified as an A2 Carcinogen by the ACGIH and as a Group 1 Carcinogen by IARC (see Toxicological Information, Section 11).

Potential Environmental Effects: Zinc concentrate will likely have minimal direct environmental effects, since its constituent metals have low solubility, and are therefore not highly bioavailable. However, when the product is processed or resides in the environment for extended periods, compound forms of zinc and lead may form which are potentially toxic to aquatic and terrestrial organisms (see Ecological Information, Section 12).

SECTION 3. COMPOSITION / INFORMATION ON INGREDIENTS

HAZARDOUS COMPONENTS	CAS Registry No.	CONCENTRATION (% wgt/wgt)
Zinc Sulphide	1314-98-3	80 to 85%
Iron Sulphide	1317-37-9	7.0 to 12%
Lead Sulphide	1314-87-0	3.0 to 5.0%
Silica	14808-60-7	2.5 to 5.0%
Cadmium Sulphide	1306-23-6	0.35 to 0.41%

Note: See Section 8 for Occupational Exposure Guidelines.

SECTION 4. FIRST AID MEASURES

Eye Contact: *Symptoms:* Eye irritation, redness. Gently brush product off face if necessary. Do not rub eye(s). Let the eye(s) water naturally for a few minutes. Look right and left, then up and down. If particle/dust does not dislodge, cautiously rinse eye(s) with lukewarm, gently flowing water for 5 minutes or until particle/dust is removed, while holding eyelid(s) open. If irritation persists, get medical advice/attention. DO NOT attempt to manually remove anything stuck to the eye.

Skin Contact: *Symptoms:* Skin soiling, mild irritation. Wash gently and thoroughly with lukewarm, gently flowing water and non-abrasive soap for 5 minutes, or until product is removed. If skin irritation occurs or you feel unwell, get medical advice/attention.

Inhalation: *Symptoms:* Respiratory irritation. Remove source of exposure or move person to fresh air and keep comfortable for breathing. Seek medical attention if you feel unwell.

Ingestion: *Symptoms:* Stomach upset. If you feel unwell or are concerned, get medical advice/attention.

SECTION 5. FIRE FIGHTING MEASURES

Fire and Explosion Hazards: Product is not considered a fire or explosion hazard. However, concentrate will burn if strongly heated in a fire situation, releasing toxic and irritating sulphur dioxide gas (SO_2). Contact with strong acids will generate flammable and highly toxic hydrogen sulphide gas (H_2S). The ignition temperature of zinc concentrate is approximately 700 – 800ºC.

Extinguishing Media: Use any means of extinction appropriate for surrounding fire conditions such as water spray, carbon dioxide, dry chemical, or foam.

Fire Fighting: Toxic fumes of sulphur dioxide will result from combustion. Fire fighters must be fully trained and wear full protective clothing including an approved, self-contained breathing apparatus which supplies a positive air pressure within a full face piece mask.

SECTION 6. ACCIDENTAL RELEASE MEASURES

Procedures for Cleanup: Control source of spillage if possible to do so safely. Restrict access to the area until completion of cleanup. Clean up spilled material immediately, observing precautions in Section 8, Personal Protection and using methods that will minimize dust generation (e.g., vacuum solids, dampen material and shovel or wet sweep). Return uncontaminated spilled material to the process if possible. Place contaminated material in suitable labeled containers for later recovery or disposal. Treat or dispose of waste material in accordance with all local, regional, and national requirements.

Personal Precautions: Persons responding to an accidental release should wear coveralls or other protective clothing, gloves and a respirator (see also Section 8). Close-fitting safety goggles may be necessary in some circumstances to prevent eye contact with dust. Workers should wash and change clothing following cleanup of a spill to prevent personal contamination with lead-containing dust.

Environmental Precautions: The handling, shipment, storage and processing of this material requires appropriate controls and care to prevent spillage or gradual accumulation in the terrestrial and aquatic environment. Spilled material should be promptly cleaned up.

SECTION 7. HANDLING AND STORAGE

Health Precautions: Some sulphide concentrates may slowly oxidize in storage and generate sulphur dioxide as well as deplete the oxygen content of a confined space. The atmosphere within confined spaces containing concentrate must be tested before entry and the area thoroughly ventilated or self-contained breathing apparatus used, if conditions warrant.

Handling (Physical Aspects): Avoid excessive heat. Avoid contact with acids, oxidizers and combustible materials. Minimize dust generation and accumulation.

Storage Precautions: Store in a cool, dry area.

Autoignition: Some sulphide concentrates may oxidize and generate heat which accumulates in storage piles. If material is to be stored for an extended period, the temperature of piles should be monitored.

Means of Control: If heating of the concentrate is detected, the material should be sealed from air or oxygen in one of the following ways:

1. Leave the piles totally intact, do not open them up or try to spread them around.
2. Tamp or compact the surface of the piles.
3. Spray the pile with water. Resort to an organic binder only if needed because it can cause formation of hard lumps and subsequent problems for processing. Suggestions for organic binders include Aerospray 70A Binder, Coherex, Igepal CA-720 and lignin sulphonate, a pulp mill by-product.
4. For smaller piles, cover them with a tarp that will prevent exposure of the material to air.
5. If inside a building or ship's hold, keep all doors closed as much as possible and minimize ventilation.

SECTION 8. EXPOSURE CONTROLS / PERSONAL PROTECTION

Occupational Exposure Guidelines:

Component	ACGIH TLV	OSHA PEL	NIOSH REL
Zinc Sulphide	None established*	None established*	None established*
Iron Sulphide	None established*	None established*	None established*
Lead Sulphide	0.05 mg Pb/m^3	0.05 mg Pb/m^3	0.05 mg Pb/m^3
Quartz (Respirable Crystalline Free Silica)	0.025 mg/m^3 Respirable SiO_2	1.4 mg/m^3 Respirable Dust** 4.3 mg/m^3 Total Dust**	0.05 mg/m^3 Respirable SiO_2
Cadmium Sulphide	0.01 mg/m^3 (Total Cd) 0.002 mg/m^3 (Respirable	0.005 mg Cd/m^3 {SECAL 0.015 / 0.05 mg Cd/m^3}	Lowest feasible level

Cd)

NOTE: OEGs for individual jurisdictions may differ from those given above. Check with local authorities for the applicable OEGs in your jurisdiction.

ACGIH - American Conference of Governmental Industrial Hygienists; OSHA - Occupational Safety and Health Administration; NIOSH - National Institute for Occupational Safety and Health. TLV – Threshold Limit Value, PEL – Permissible Exposure Limit, REL – Recommended Exposure Limit.

* - NOTE: While there are no established OELs for zinc sulphide and iron sulphide as such, there are OELs for their respective oxides, which may be formed during burning, welding or other fuming processes. The OSHA PEL for zinc oxide dust is 15 mg/m^3 (total) and 5 mg/m^3 (respirable); the OSHA PEL for zinc oxide fume is 5 mg/m^3. The ACGIH TLV for zinc oxide is 2 mg/ m^3 (respirable fraction) with a Short Term Exposure Limit (STEL) of 10 mg/m^3 (respirable fraction). The NIOSH REL for zinc oxide (dust or fume) is 5 mg/m^3 10 hr TWA with a 15 mg/m^3 ceiling for zinc oxide dust and a 10 mg/m^3 STEL for zinc oxide fume (15 min. sample).
The OSHA PEL for iron oxide fume is 10 mg/m^3. The NIOSH REL for iron oxide dust and fume is 5 mg/m^3 (as Fe) and the ACGIH TLV is 5 mg/m^3 of iron oxide dust/fume (respirable fraction).
** - NOTE: The OSHA PEL for silica applies to the total airborne zinc concentrate dust concentration and has been calculated based on the maximum percent SiO_2 in the sample using the formulas: Respirable Dust PEL = 10 $mg/m^3/(\%SiO_2 + 2)$: Total Dust PEL. = 30 $mg/m^3/(\%SiO_2 + 2)$ (see Table Z-3 of 29 CFR 1910.1000).
SECAL: To be achieved in specified processes and work places where it is not possible to achieve the PEL through engineering and work practices alone. The OSHA SECAL for cadmium is 0.015 or 0.05 mg/m^3, depending on the processes involved. See Table 1 of 29 CFR 1910.1027.

NOTE: The selection of the necessary level of engineering controls and personal protective equipment will vary depending upon the conditions of use and the potential for exposure. The following are therefore only general guidelines that may not fit all circumstances. Control measures to consider include:

Ventilation: Use adequate local or general ventilation to maintain the concentration of zinc concentrate dust in the working environment well below the appropriate occupational exposure limits. Supply sufficient replacement air to make up for air removed by the exhaust system.

Protective Clothing: Coveralls or other work clothing, glasses or goggles, and gloves are recommended to prevent prolonged or repeated direct skin contact. Close-fitting safety goggles should be worn to prevent eye contact if excessive dust is generated. Workers should wash immediately when skin becomes contaminated and at the end of each work shift. Work clothing should be removed immediately if it becomes heavily contaminated and should be changed daily and laundered before reuse if there is a reasonable probability that the clothing may be contaminated.

Respirators: Where zinc concentrate dust is generated and cannot be controlled to within acceptable levels by engineering means, use appropriate NIOSH-approved respiratory protection equipment (a 42CFR84 Class N, R or P-100 particulate filter cartridge).

General Hygiene Considerations: Avoid breathing dust. Always practice good personal hygiene. Refrain from eating, drinking, or smoking in work areas. Thoroughly wash hands after handling and before eating, drinking, or smoking in appropriate designated areas only.

SECTION 9. PHYSICAL AND CHEMICAL PROPERTIES

Appearance: Dark grey-brown, fine-grained powder	**Odour:** Weak organic odour from entrained flotation reagents	**Odour Threshold:** No data	**pH:** Not applicable
Vapour Pressure: Negligible at 20°C	**Vapour Density:** Not Applicable	**Melting Point/Range:** Will burn first unless in an inert atmosphere	**Boiling Point/Range:** Not Applicable
Relative Density (Water = 1)**:** 2.0 (Bulk Sp. Gr.)	**Evaporation Rate:** Not Applicable	**Coefficient of Water/Oil Distribution:** Not Applicable	**Solubility:** Essentially Insoluble
Flash Point: Not Applicable	**Flammable Limits (LEL/UEL):** Non Flammable	**Auto-ignition Temperature:** None	**Decomposition Temperature:** >1000°C
Particle Size: <40 μm, with 80% <20 μm	**Percent Volatiles:** 8.4% @ 100°C (moisture)		

SECTION 10. STABILITY AND REACTIVITY

Stability & Reactivity: Material is stable and not considered reactive under normal temperatures and pressures. Hazardous polymerization or runaway reactions will not occur.

Incompatibilities: Reacts violently with iodine pentachloride. Incompatible with iodine monochloride, hydrogen peroxide, strong oxidizers, and strong acids.

Hazardous Decomposition Products: May release highly toxic and flammable hydrogen sulfide gas on contact with strong acids. This material can decompose at high temperatures forming toxic and irritating sulphur dioxide gas as well as zinc, lead and cadmium oxides.

SECTION 11. TOXICOLOGICAL INFORMATION

General: In the powder form in which this product is sold, the metals are present as sulphides that are relatively insoluble and poorly absorbed within the body. However, high temperature operations such as oxy-acetylene cutting, electric arc welding or arc-air gouging on dust–contaminated surfaces will generate zinc oxide fume that also contains lead and cadmium oxides. These oxides are soluble in body fluids and the particle size of the metal fumes is largely within the respirable size range, which increases the likelihood of inhalation and deposition of the fume within the body. The primary route of exposure would be through inhalation of metal oxide fumes, composed principally of zinc oxide and including some lead and cadmium oxides.

NOTE: The toxicological properties of this material have not been fully investigated. The information contained in this SDS is therefore based on information in the technical and scientific literature about the material's constituent compounds.

Acute: Skin/Eye: Contact with dust or fume may cause local irritation but would not cause tissue damage.

Inhalation: Exposure to dust or fume is irritating to the nose, throat and respiratory tract with dryness and irritation of the nose and throat, possible tightness of the chest, coughing and metallic taste. It may cause headache, as well as gastrointestinal disturbances. An intense, short-term exposure to welding/burning fumes could cause congestion and pulmonary edema. However, short-term exposures of this magnitude are unlikely in industry today. Less intense short-term exposure could result in the condition called metal fume fever. The symptoms of metal fume fever will occur within 3 to 10 hours, and include immediate dryness and irritation of the throat, tightness of the chest, and coughing which may later be followed by flu-like symptoms of fever, malaise, perspiration, frontal headache, muscle cramps, low back pain, occasionally blurred vision, nausea, and vomiting. The symptoms are temporary and generally disappear, without medical intervention, within 24 to 48 hours of onset. There are no recognized complications, after affects, or chronic affects that result from zinc metal fume fever. An acute, short-term exposure to high levels of oxide fumes could also result in the absorption of lead and cadmium in the body. Kidney damage, as well as anemia, could then result from acute exposure.

Ingestion: Symptoms due to ingestion of dust or fume would be similar to those from inhalation. Other health effects such as constipation or bloody diarrhea might also occur.

Chronic: The chronic health effects of zinc concentrate have not been fully investigated. Prolonged exposure to zinc concentrate dust may be expected to produce many of the symptoms of short-term exposure and may also cause central nervous system damage, gastrointestinal disturbances, kidney dysfunction, anemia, and possible skin rashes or dermatitis. Reduced hemoglobin production has been associated with low lead exposures. Symptoms of central nervous system damage due to moderate lead exposure include fatigue, headaches, tremors and hypertension. Very high exposure can result in lead encephalopathy with symptoms of hallucinations, convulsions, and delirium. Kidney dysfunction and possible injury has also been associated with chronic lead and cadmium poisoning. Chronic over-exposure to lead has been implicated as a causative agent for the impairment of male and female reproductive capacity. Pregnant women should be protected from excessive exposure as lead can cross the placental barrier and unborn children may suffer neurological damage or developmental problems. Teratogenic and mutagenic effects from exposure to lead have been reported in some studies but not in others. The literature is inconsistent and no firm conclusions can be drawn at this time. Lead and lead compounds are listed as an *A3 Carcinogen (Confirmed Animal Carcinogen with Unknown Relevance to Humans)* by the ACGIH. IARC has listed lead compounds as *Group 2A Carcinogens (Probably Carcinogenic to Humans)*. The NTP has listed lead and lead compounds as *Reasonably Anticipated to be a Human Carcinogen*. OSHA and the EU do not currently list lead as a human carcinogen. IARC has classified cadmium and certain cadmium compounds as a *Group 1 Carcinogen (Carcinogenic to Humans)* while ACGIH classifies cadmium as a *Suspected Human Carcinogen (A2)*. The NTP classifies cadmium as a *Known Human Carcinogen* and OSHA lists cadmium as a *Carcinogen.* The European Union (EU) classifies cadmium sulphide as a *Category 3 (Possible) Carcinogen*. IARC has classified crystalline silica of respirable particle size as a *Group 1 Carcinogen (Carcinogenic to Humans)* while ACGIH classifies it as a *Suspected Human Carcinogen (A2)*. The NTP classifies silica as a *Known Human Carcinogen.* OSHA and the EU do not list silica as a carcinogen.

Animal Toxicity:

Hazardous Ingredient:	Acute Oral Toxicity:	Acute Dermal Toxicity:	Acute Inhalation Toxicity:
Zinc Sulphide	>2,000 mg/kg†	>2000 mg/kg*	>5.04 mg/L/4hr‡
Iron Sulphide	no data	no data	no data
Lead Sulphide	no data	no data	no data
Silica	no data	no data	no data
Cadmium Sulphide	7,080 mg/kg†	no data	no data

† LD_{50}, Rat,Oral, * LD_{50}, Rat, Dermal ‡ LC_{50}, Rat, Inhalation, 4 hour

SECTION 12. ECOLOGICAL INFORMATION

Zinc concentrate will likely have minimal direct environmental effects, since its constituent metals have low solubility, and are therefore not highly bioavailable. However, when the product is processed or resides in the environment for extended periods, compound forms of zinc and lead may form which are potentially toxic to aquatic and terrestrial organisms. The mobilities of zinc and lead are media-dependent. These elements can bind with inorganic and organic ligands, reducing their mobility and bioavailability in soil and water. Bioavailability is also regulated by other physico-chemical factors, such as pH and hardness.

Zinc: Zinc is potentially toxic to aquatic organisms. In aquatic systems, zinc bioaccumulates in both plants and animals. Zinc also bioaccumulates in terrestrial plants, vertebrates, and mammals; plant uptake from soil depends upon the plant species, soil pH, and soil composition. In general, zinc does not biomagnify through food chains.

Lead: Lead compounds are potentially persistent in the aquatic environment. Dissolved lead compounds have the potential to bioaccumulate in plants and animals, both aquatic and terrestrial. Lead may occur as sorbed ions or surface coatings on sediment mineral particles, or may be carried in colloidal particles in surface water. Most lead is strongly retained in soil, resulting in relatively low mobility. Lead may be immobilized by ion exchange with hydrous oxides or clays or by chelation with humic or fulvic acids in the soil.

SECTION 13. DISPOSAL CONSIDERATIONS

If material cannot be returned to process or salvage, dispose of only in accordance with applicable regulations. Spilled concentrate may meet the requirements of a hazardous waste in most jurisdictions. It is the responsibility of the waste generator to determine the toxicity and physical properties of the material generated in order to determine the proper waste classification and disposal methods.

SECTION 14. TRANSPORT INFORMATION

TRANSPORT CANADA CLASSIFICATION ..... Not regulated
U.S. DOT HAZARD CLASSIFICATION ..... Class 9, Packing Group III
U.S. PROPER SHIPPING NAME ..... Environmentally Hazardous Substance, Solid, n.o.s. (contains lead sulphide)
U.S. DOT RQ ..... Lead sulphide 10 lbs.
U.S. DOT PRODUCT IDENTIFICATION NUMBER UN3077
MARINE POLLUTANT ..... No
IMO CLASSIFICATION ..... MHB - Materials Hazardous Only in Bulk, Group A and B

Note that this material has been tested under the United Nations Transport of Dangerous Goods, Manual of Tests and Criteria, Fifth Revised Edition (2009). Test results indicate that the concentrate qualifies neither as a flammable solid under Class 4.1 nor a self-heating substance under Class 4.2.

Risks: This material may liquefy if shipped at moisture content in excess of its transportable moisture limit (TML). It may also present chemical hazards. Recommendations set out in Appendix 1 of the International Marine Solid Bulk Cargo Code should be observed.

IMO MARPOL V Classification: Not Harmful to the Marine Environment.

SECTION 15. REGULATORY INFORMATION

U.S.

INGREDIENTS LISTED ON TSCA INVENTORY ..... Yes

HAZARDOUS UNDER HAZARD COMMUNICATION STANDARD ..... Lead Sulphide .....Yes
Cadmium Sulphide .....Yes

	Silica .....Yes
CERCLA SECTION 103 HAZARDOUS SUBSTANCES	Lead Sulphide.....RQ: 10 lbs. (4.54 kg.) Zinc Compounds .....RQ: None assigned Cadmium Compounds.....RQ: None assigned
EPCRA SECTION 302 - EXTREMELY HAZARDOUS SUBSTANCE.....	None of the ingredients qualify.
EPCRA SECTION 311/312 HAZARD CATEGORIES	Delayed (chronic) Health Hazard - Carcinogen Delayed (chronic) Health Hazard – Reproductive Toxin
EPCRA SECTION 313 TOXIC RELEASE INVENTORY.....	Lead Compounds (Lead Sulphide) CAS No 1314-87-0 Percent by Weight: .....3.0% to 5.0% Zinc Compounds (Zinc Sulphide) CAS No 1314-98-3 Percent by Weight: .....80% to 85% Cadmium Compounds (Cadmium Sulphide) CAS No 1306-23-6 Percent by Weight .....0.35% to 0.41%

SECTION 16. OTHER INFORMATION

Date of Original Issue: September 24, 1997 **Version:** 01 *(First edition)*

Date of Latest Revision: June 12, 2015 **Version:** 15

The information in this Safety Data Sheet is based on the following references:

- American Conference of Governmental Industrial Hygienists, 2004, Documentation of the Threshold Limit Values and Biological Exposure Indices, Seventh Edition plus updates.
- American Conference of Governmental Industrial Hygienists, 2015, Guide to Occupational Exposure Values.
- American Conference of Governmental Industrial Hygienists, Threshold Limit Values for Chemical Substances and Physical Agents and Biological Exposure Indices – 2015.
- Bretherick's Handbook of Reactive Chemical Hazards, 20th Anniversary Edition. (P.G. Urben Ed.) 1995.
- Canadian Centre for Occupational Health and Safety (CCOHS), Hamilton, ON, CHEMINFO Record No. 608 Lead.
- Canadian Centre for Occupational Health and Safety (CCOHS), Hamilton, ON, CHEMINFO Record No. 548 Zinc.
- Canadian Centre for Occupational Health and Safety (CCOHS), Hamilton, ON, CHEMINFO Record No. 3454 Cadmium.
- European Economic Community, Commission Directives 91/155/EEC and 67/548/EEC.
- Industry Canada, SOR/2015-17, January 30, 2015 - Hazardous Products Regulations.
- International Agency for Research on Cancer (IARC), Monographs on the Evaluation of the Carcinogenic Risk of Chemicals to Man, 1972 – present, (multi-volume work), World Health Organization, Geneva.
- International Chemical Safety Cards (WHO/IPCS/ILO), ICSC:0052 – Lead, ICSC 1627 – Zinc Sulphide, ICSC 0404 – Cadmium Sulphide, ICSC 0808 – Quartz.
- Merck & Co., Inc., 2001, The Merck Index, An Encyclopedia of Chemicals, Drugs, and Biologicals, 13th Edition.
- National Library of Medicine, National Toxicology Information Program, Hazardous Substance Data Bank (on-line edition).
- Patty's Toxicology, 5th Edition, (E Bingham, B Cohrssen & C H Powell, Ed.) 2001.
- U.S. Dept. of Health and Human Services, National Institute of Environmental Health Sciences, National Toxicology Program (NTP), 13th Report on Carcinogens, October 2014.
- U.S. Dept. of Health and Human Services, National Institute for Occupational Safety and Health, NIOSH Pocket Guide to Chemical Hazards. CD-ROM Edition September 2005.
- U.S. Dept. of Health and Human Services, National Institute for Occupational Safety and Health, Registry of Toxic Effects of Chemical Substances (RTECS).
- U.S. Dept. of Health and Human Services, Public Health Service, Agency for Toxic Substances and Disease Registry, Toxicological Profile for Lead.
- U.S. Dept. of Health and Human Services, Public Health Service, Agency for Toxic Substances and Disease Registry, Toxicological Profile for Zinc.
- U.S. Dept. of Health and Human Services, Public Health Service, Agency for Toxic Substances and Disease Registry, Toxicological Profile for Cadmium.
- U.S. Occupational Safety and Health Administration, 1989, Code of Federal Regulations, Title 29, Part 1910.

Notice to Reader

Although reasonable precautions have been taken in the preparation of the data contained herein, it is offered solely for your information, consideration and investigation. Teck Alaska Incorporated extends no warranty and assumes no responsibility for the accuracy of the content and expressly disclaims all liability for reliance thereon. This safety data sheet provides guidelines for the safe handling and processing of this product; it does not and cannot advise on all possible situations. Therefore, your specific use of this product should be evaluated to determine if additional precautions are required. Individuals exposed to this product should read and understand this information and be provided pertinent training prior to working with this product.